

Quantifying iron fertilization with stable isotopes



**Non-traditional Isotope Research on
Various Advanced Novel Applications**

ExOIS Forum

2022-08-15

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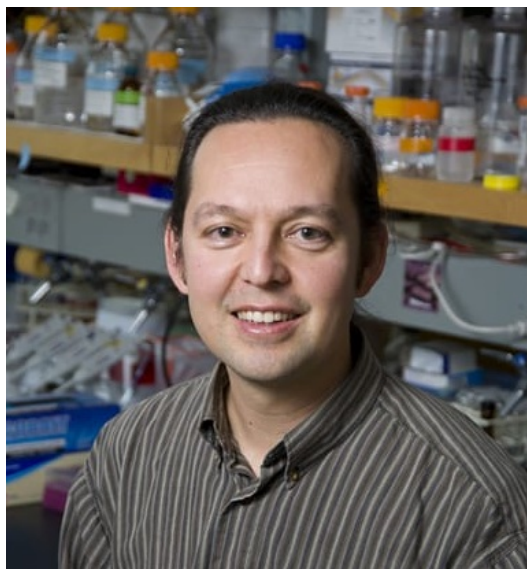
The team, supported by *OCIA*



Tristan Horner

WHOI MC&G

(person talking now)



Mak Saito

WHOI MC&G



Ichiko Sugiyama

Weizmann Inst.

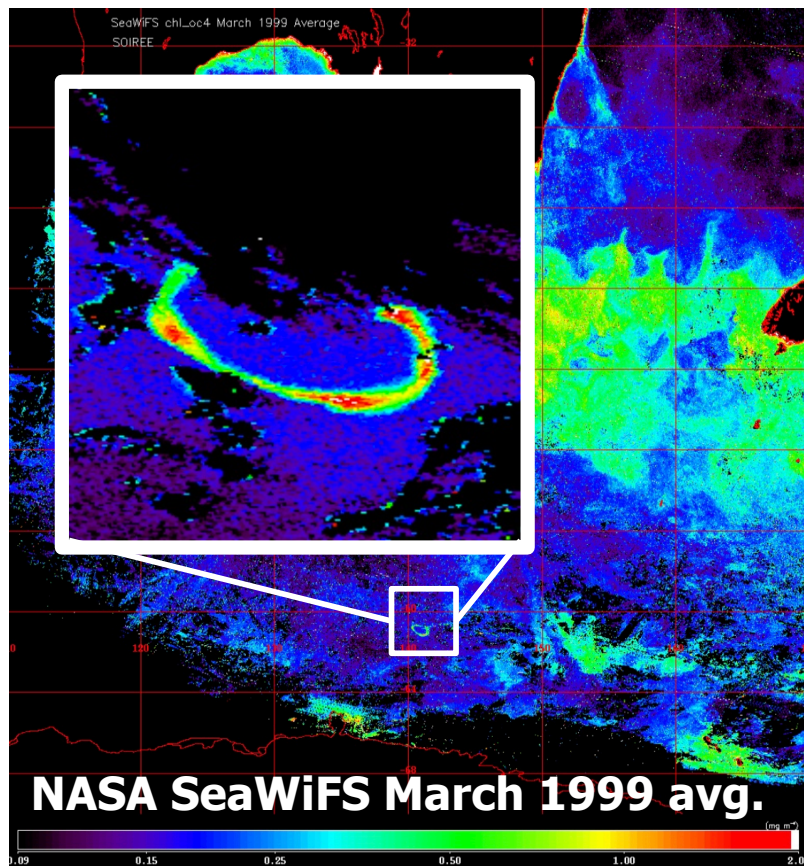
Soon: WHOI MC&G

The three uncertainty's of fertilization

Effect. Does fertilization stimulate carbon export?

Efficiency. What is the best way to add Fe?

Efficacy. Do plankton encounter a secondary nutrient limitation?

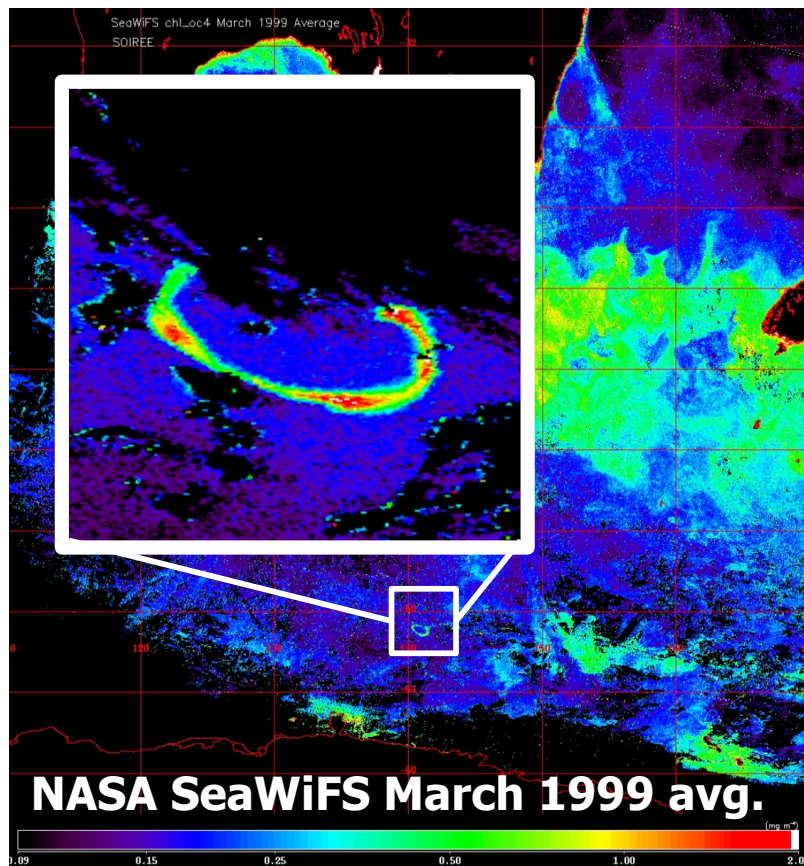


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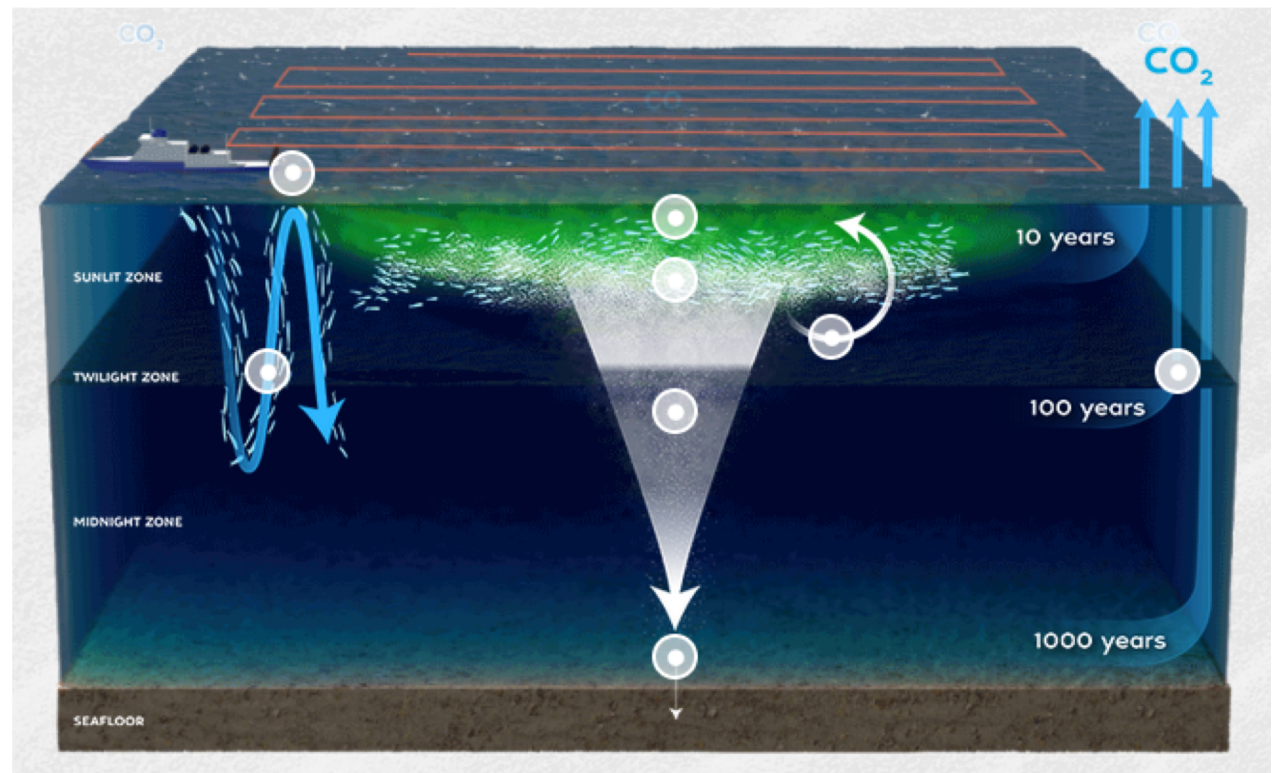
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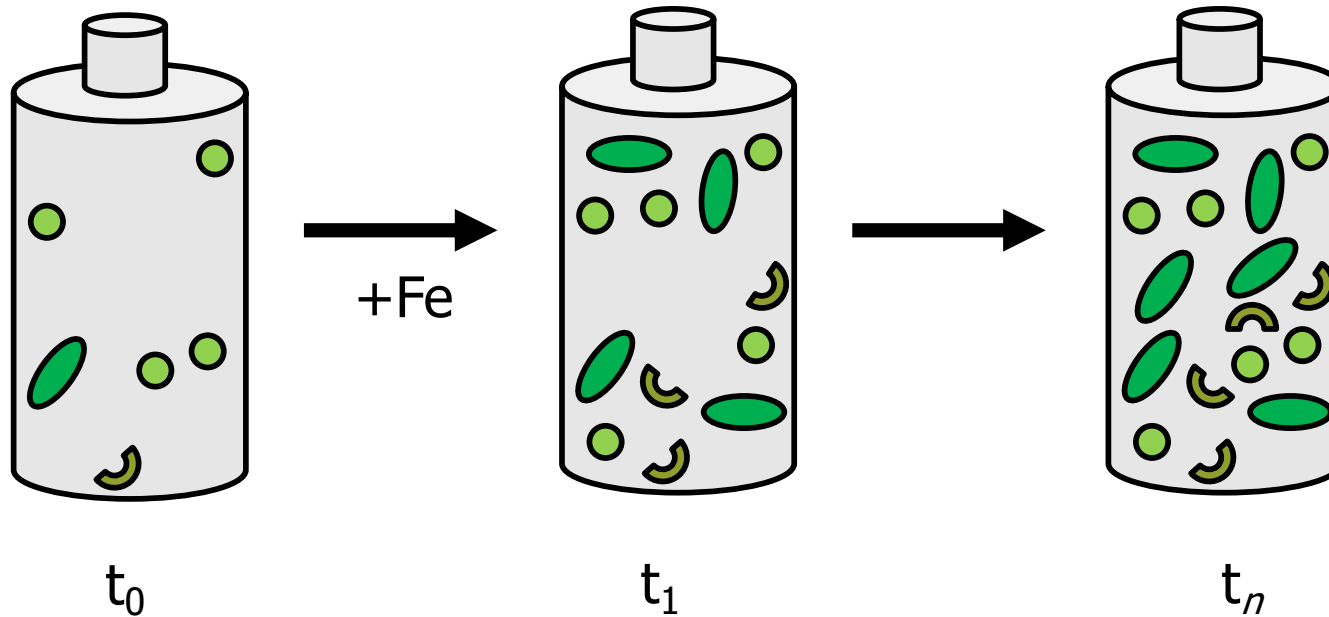


Trust, but verify

How can we **verify** that observed biomass was formed in response to fertilization?

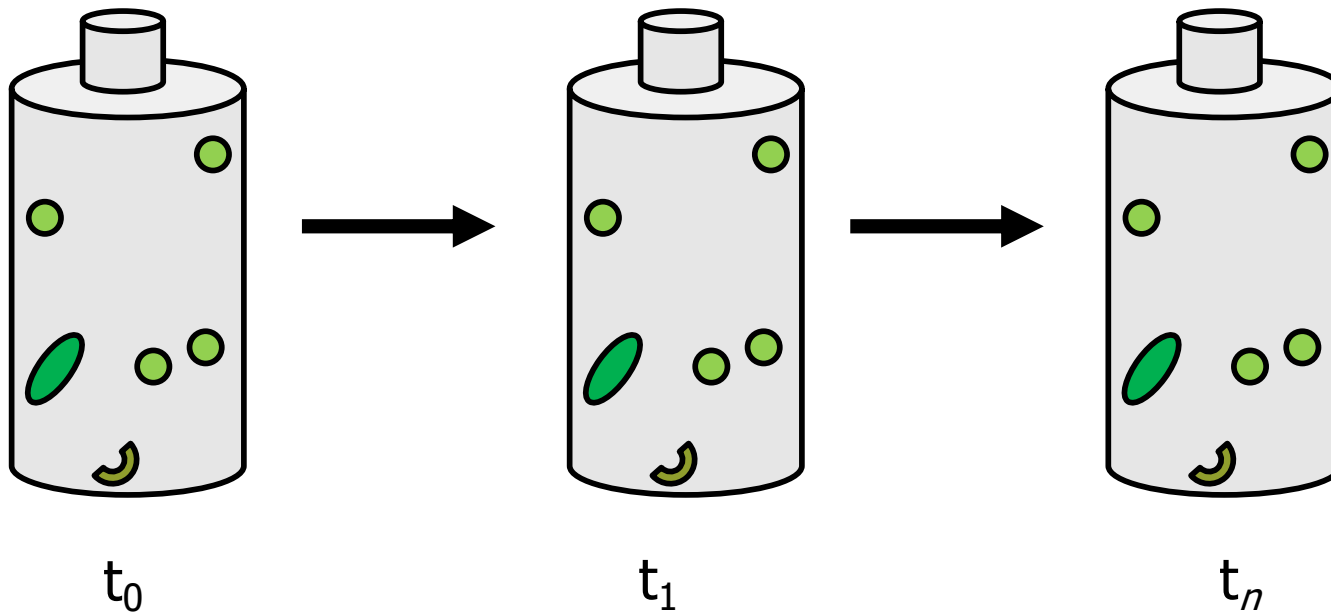


Verifying effects in enclosed experiments



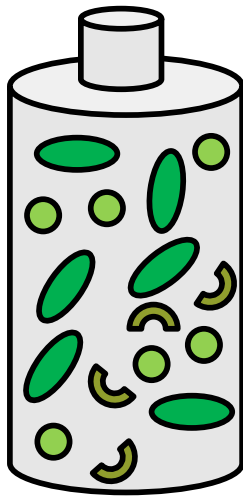
Amended

Verifying effects in enclosed experiments



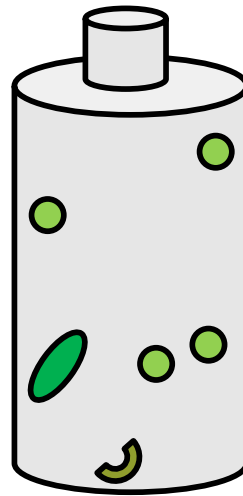
Unamended

Verifying effects in enclosed experiments



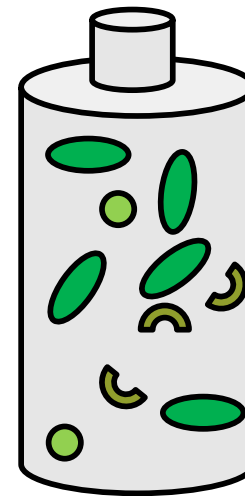
Amended

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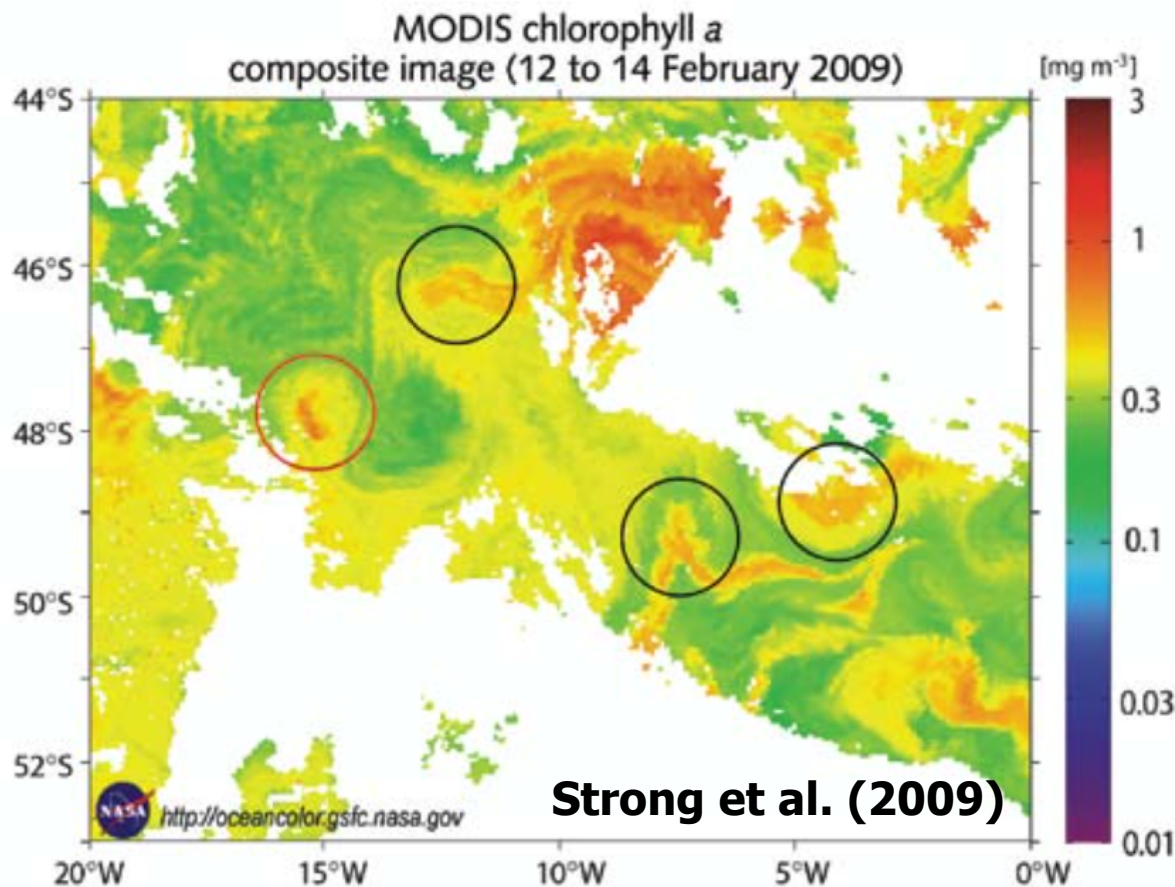
Unamended

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Growth in response
to fertilization

How do we put the ocean in a bottle?



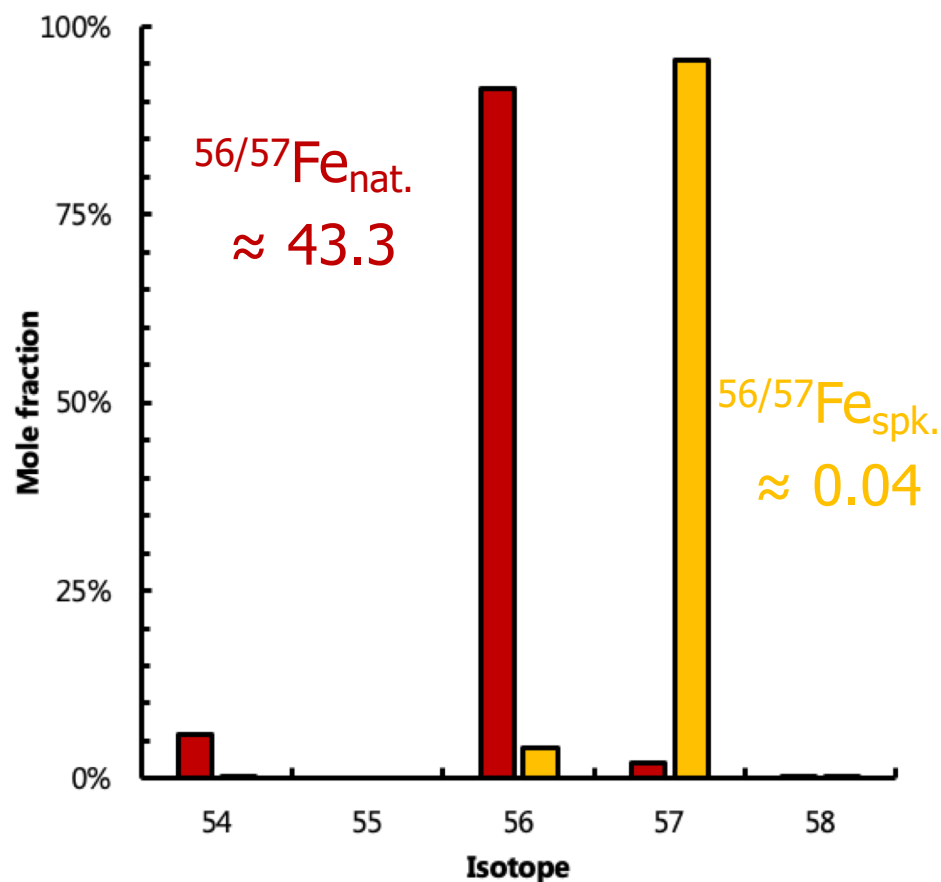
“ocean fertilization ... would have to demonstrate that the iron-induced bloom would not have otherwise occurred”

The idea

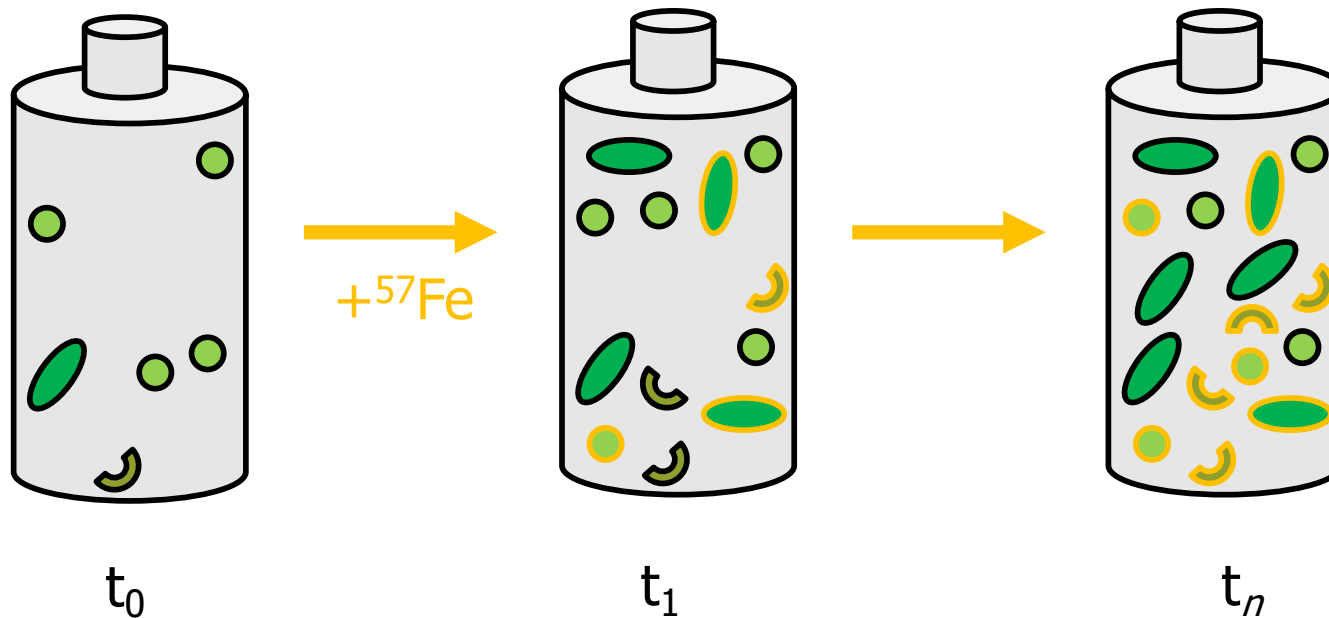
Can we use iron-57 to address the three E's?

Iron has four stable isotopes; most is iron-56

Can purchase enriched **spikes** of other isotopes



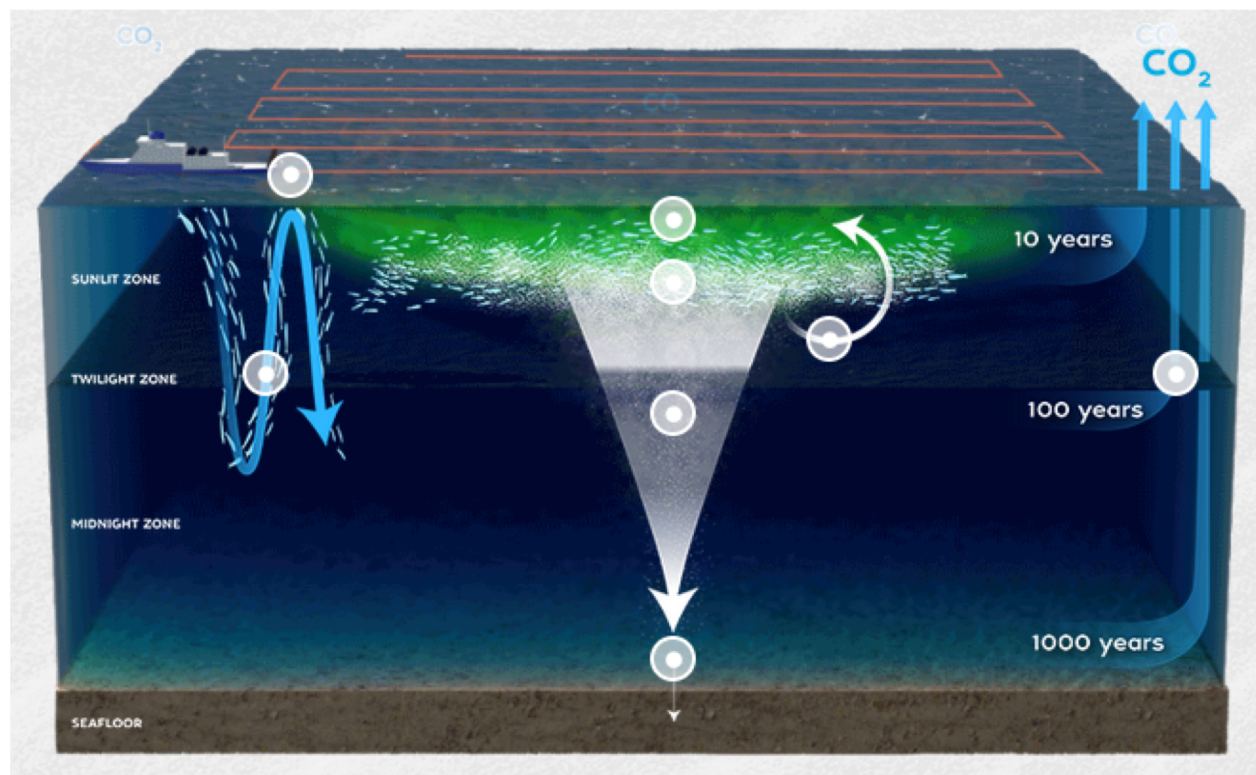
Verification using iron-57



$^{56}\text{Fe}:^{57}\text{Fe}$ of biomass diagnostic of the iron source—fertilized or not

Verifying in unenclosed experiments

Iron isotope ratio of sinking particles will tell you if export from was Fe fertilization (low $^{56}\text{Fe}:^{57}\text{Fe}$) or not (normal $^{56}\text{Fe}:^{57}\text{Fe}$)

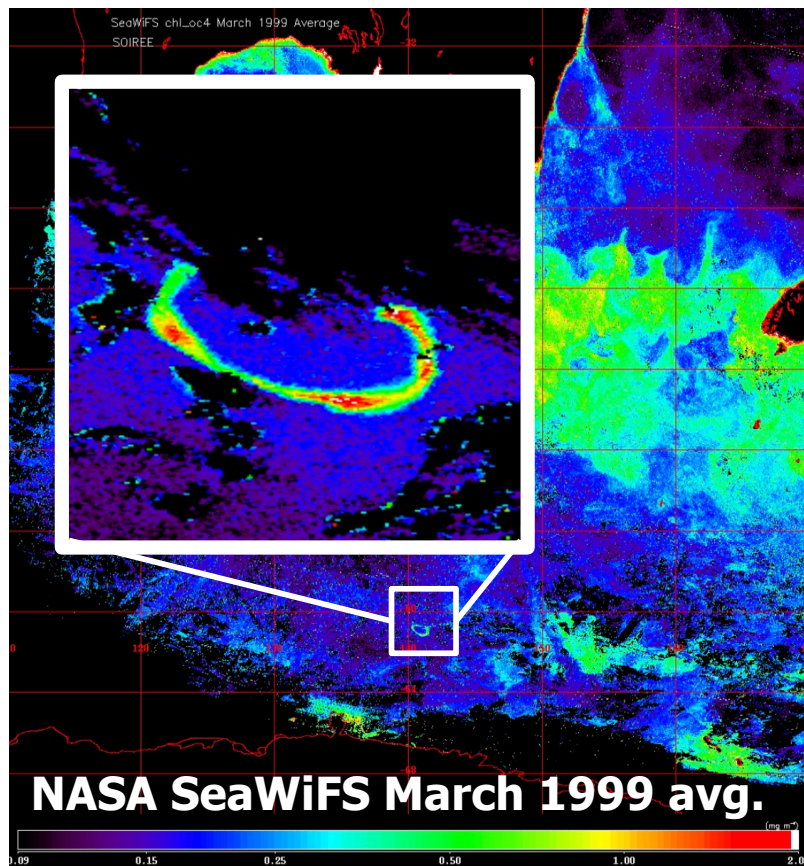


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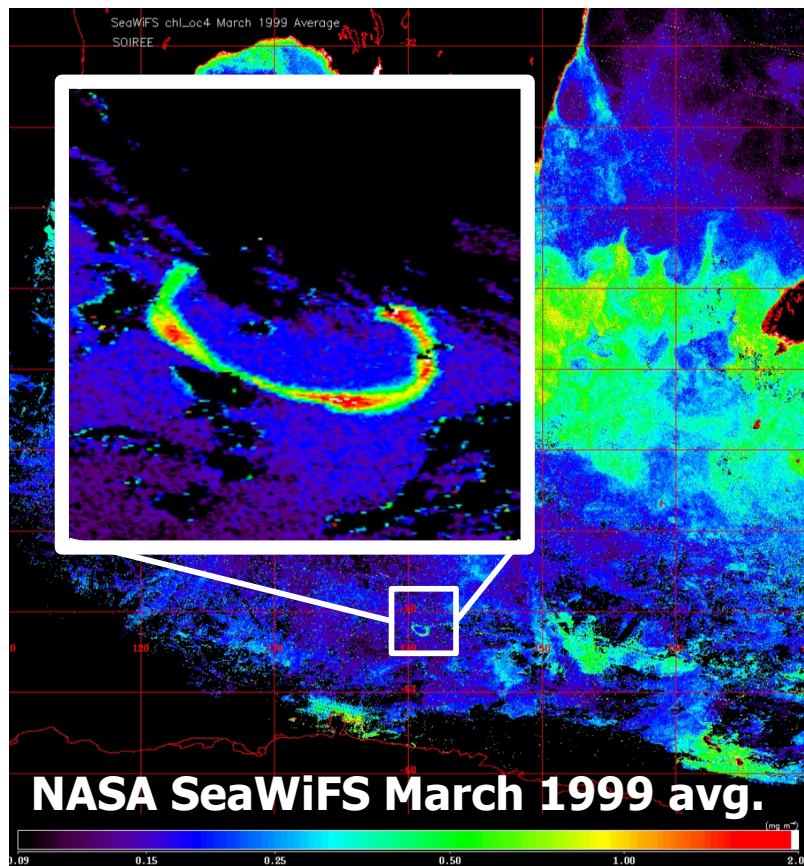


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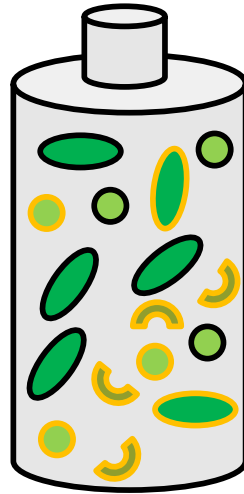
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Fertilizing efficiently

How much Fe
dissolves relative
to how much Fe is
added?

How much iron is
taken up by cells
relative to the
bottle walls?



- Ligand
 - Sulfate
 - Chloride
 - ...?
- Quantity
 - Amount over background
- Rate
 - All at once?
 - Every day?

Fertilizing efficiently

Cost of iron

- Normally ~\$0.01 per g
- Iron-57 ~\$5,000 per g
- Need to consider ways to bring cost down if scaling up

Areas for improvement

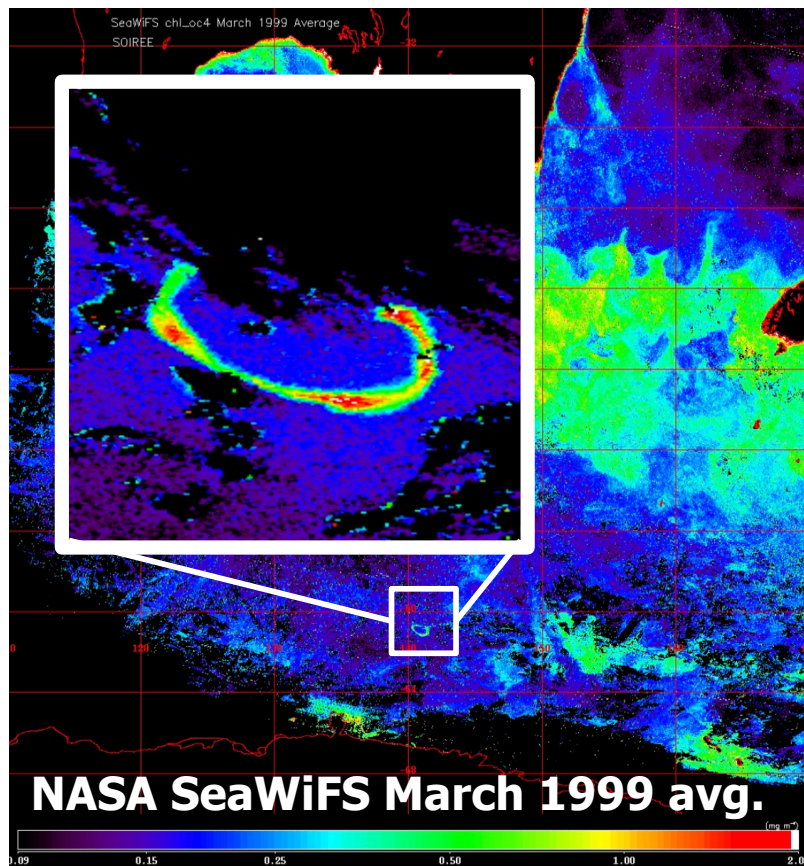
- Not all Fe need come from ^{57}Fe , just enough to measure
- Improve measurement precision (use MC-ICP-MS)
- Fertilize certain regions with ^{57}Fe , use ^{56}Fe elsewhere
- Lower iron-57 production costs (dedicated cyclotrons?)

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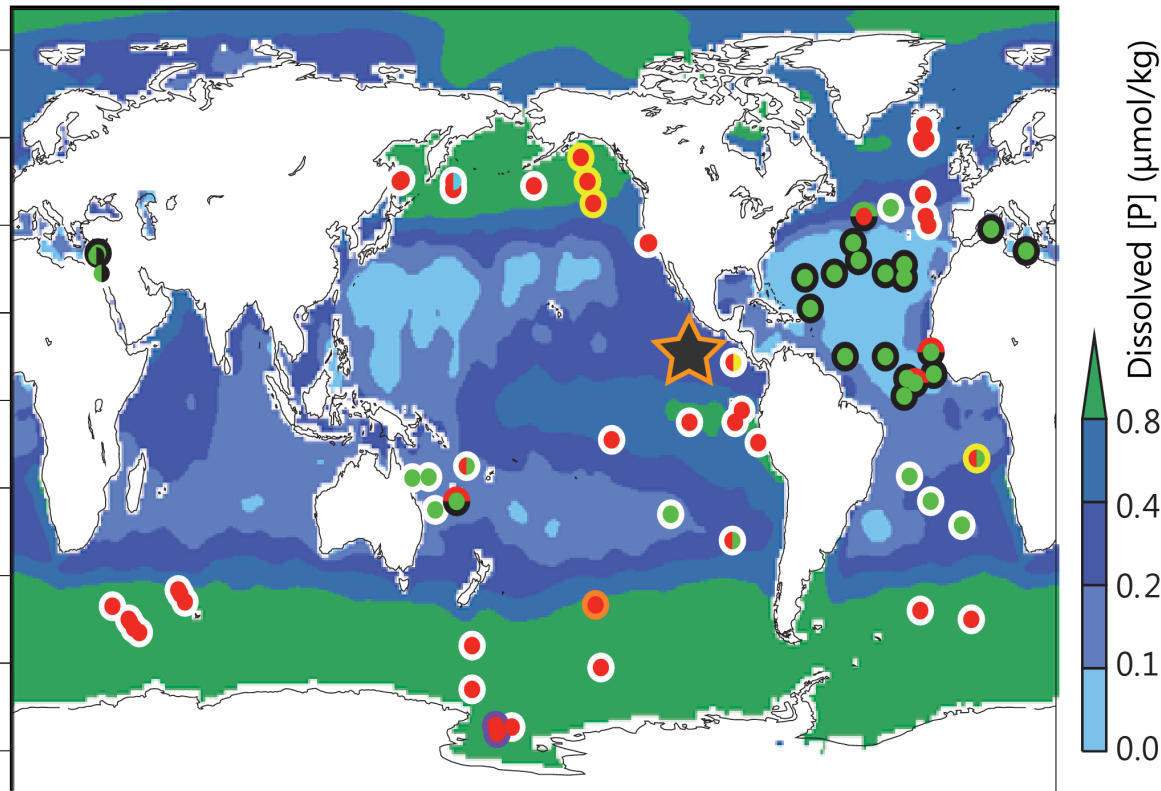
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Maximizing efficacy



Moore et al. (2013)

Site-specific considerations:

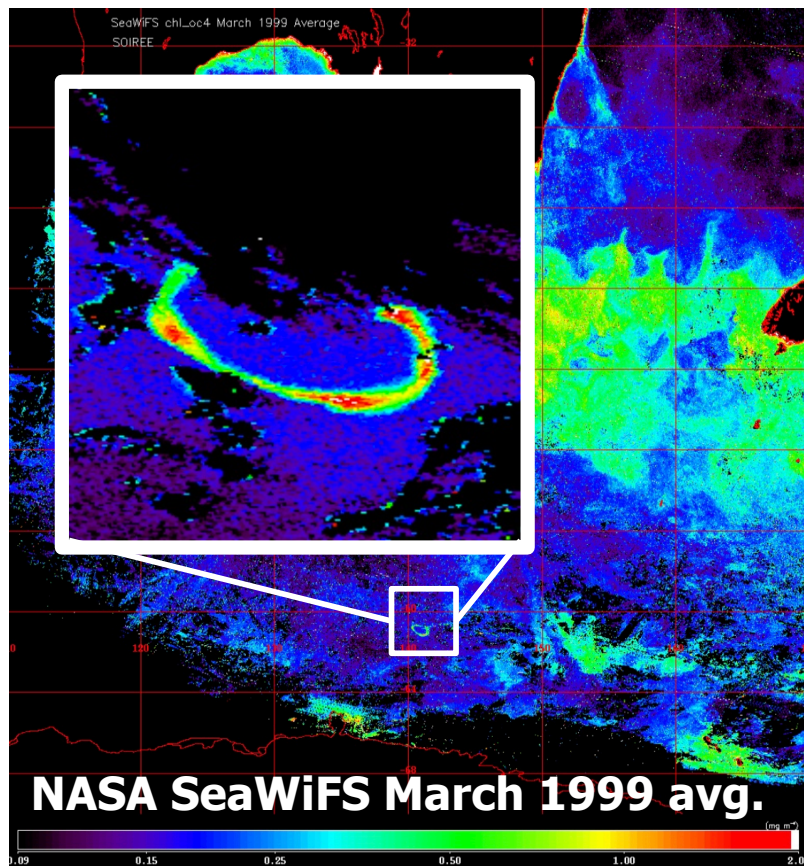
- Do cells increase Fe quotas?
- Will ecology quickly respond?
- Are there secondary nutrient limitations?

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